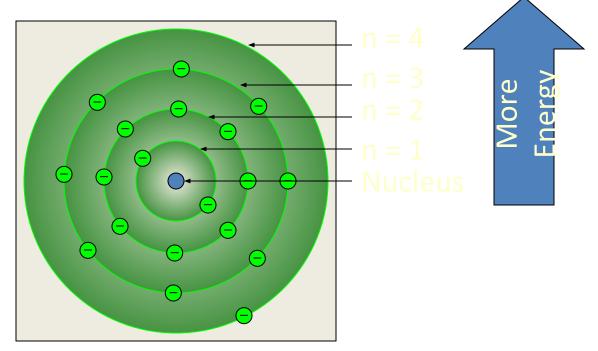
# Electron Configuration

#### **Electron Structure**

- In a **neutral** atom, electrons = atomic number.
- Electrons are arranged into energy levels.
- Energy Level (shell) defines how far the electron is away from the nucleus.



Simple model of a potassium atom

## **Electron Arrangement**

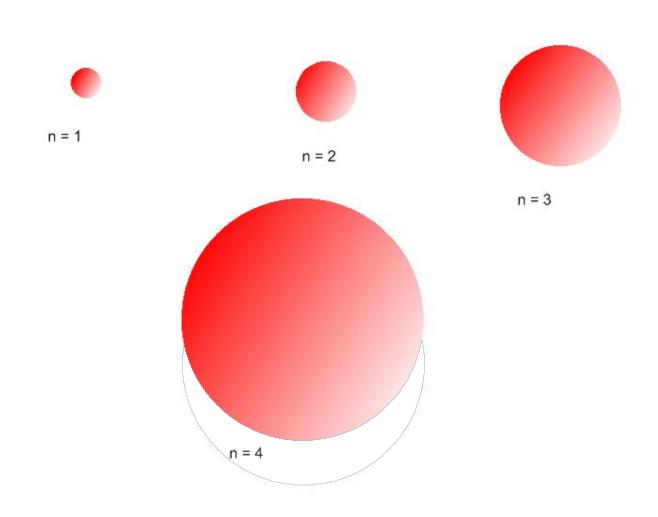
Aufbau Rule: electrons fill into the lower energy orbitals before moving to higher energy orbitals

Hund's Rule: one electron must be in each position of an orbital before they are paired

Pauli exclusion principle: no two electrons can have the same set of quantum numbers (define properties of e<sup>-</sup>)

• Principle Quantum number: Energy Level (shell) defines how far the electron is away from the nucleus.

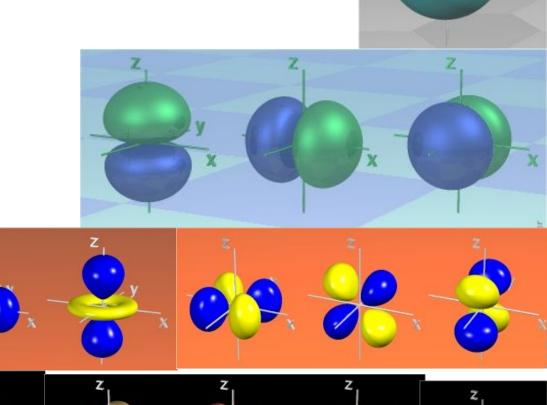
#### **Bohr Model rings**



## **Orbitals**

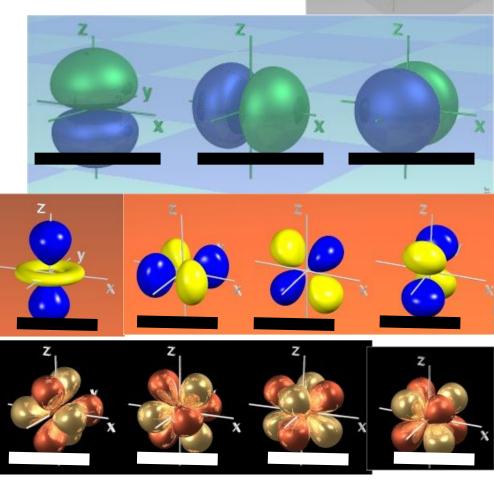
• Orbital quantum number: defines the shape of orbital

- –Four types:
  - S
  - p
  - d
  - f



Magnetic quantum number: defines the orientation in space of the orbital (number of available spaces (positions) for electrons to fill

- Two electrons can fill each position
- How many electrons can each orbital hold?



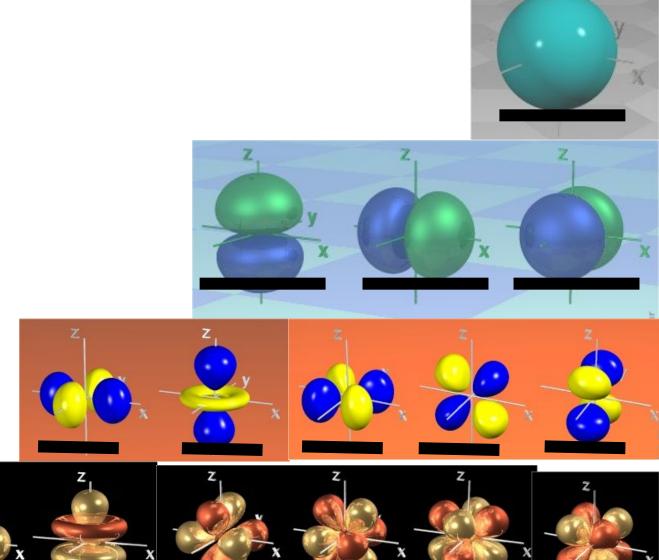
### **Electron Spin**

<u>Spin Quantum number:</u> designates direction of electron spin (must be spinning in opposite directions).

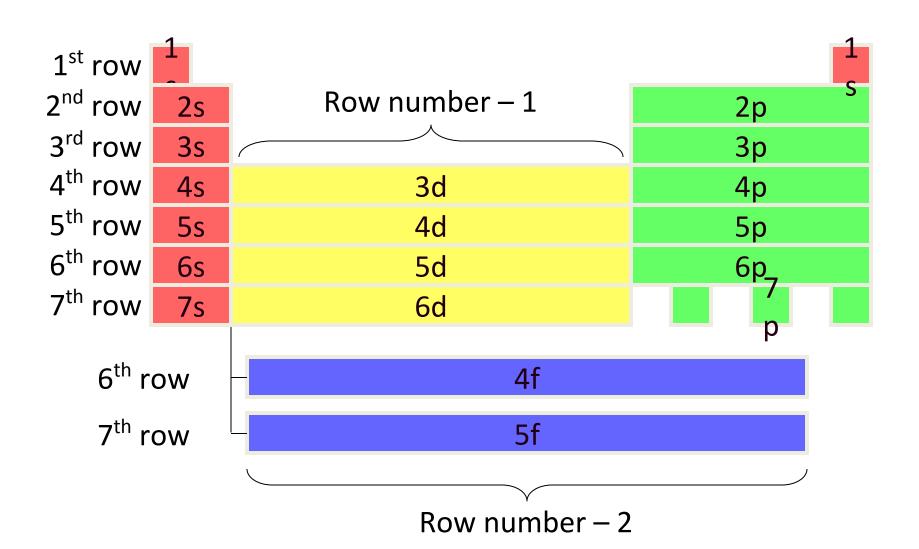
- Each orbital can hold up to 2 electrons.
- Why only 2 in each orbital?
  - Spin up
  - Spin down

#### **Electron Structure**

- Orbitals
  - -s/1 position  $/2e^{-}$
  - -p/3 positions  $/6e^{-}$
  - -d/5 positions/10 e<sup>-</sup>
  - -f / 7 positions / 14 e<sup>-</sup>



## **Electron Configurations**



#### **Orbital Notations**

- Combines electron configuration with the spin quantum number.
- Use the configurations to help you write the correct orbital notation.
- Don't forget the rules:
  - Aufbau
  - Hund
  - Pauli exclusion

### Noble Gas Configuration

- Find your element.
- Find the noble gas (Group 8A) before your element.
- -Write the noble gas in brackets.
- Add electrons to the noble gas until you reach your element.